

B.Pharm II Year II Semester (R15) Regular Examinations May/June 2017

**PHARMACEUTICAL ANALYSIS – I**

Time: 3 hours

Max. Marks: 70

**PART – A**

(Compulsory Question)

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- 1 Answer the following: (10 X 02 = 20 Marks)
- Define significant figure.
  - Define Lewis acid and base with example.
  - Write indicators used in complexometric titration.
  - Define adsorption indicator with example.
  - Give Nernst equation with notations.
  - Define specific conductance.
  - Define resonance, fluorescence and Stokes shift.
  - Enlist different atomizers used in AAS.
  - Define unpolarised and plane polarised light.
  - Enlist the main parts of polarimeter.

**PART – B**

(Answer all five units, 5 X 10 = 50 Marks)

**UNIT – I**

- 2 (a) Explain different types of errors with examples.  
(b) Explain the various methods to minimize the errors.

**OR**

- 3 (a) Define solubility product and explain its effects.  
(b) Explain the principle and reaction involved in the assay of sodium benzoate.

**UNIT – II**

- 4 (a) Describe the different types of complexometric titration with suitable examples.  
(b) Explain the principle involved in the complexometric titration with suitable example.

**OR**

- 5 (a) Describe permanganometry and iodimetry with reactions.  
(b) Explain the principle and reaction involved in the assay of copper sulphate.

**UNIT – III**

- 6 Explain the various methods used to determine the end point in potentiometry with graph.

**OR**

- 7 With neat labelled diagram, explain the construction, working, advantages and disadvantages of dropping mercury electrode.

**UNIT – IV**

- 8 Describe the principle, instrumentation and applications of Atomic Absorption Spectrophotometer.

**OR**

- 9 Explain the spectral and chemical interferences that occur in Atomic Absorption Spectroscopy.

**UNIT – V**

- 10 Describe various grades of reagents used in QC lab and discuss the applications of L.R grade, A.R Grade and 1-LPLC grade reagents in Pharmaceutical industry.

**OR**

- 11 How moisture content will be determined in Pharmaceutical industries and discuss in detail Karl-Fisher method with neat labelled diagram.

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